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observed dark color of the crystals may indicate some transfer of charge. The r.m.s. deviation of C atoms from the plane of the anthracene ring system is 0.003 (1) Å, while for the dianhydride it is 0.027 Å. Atoms O(12), O(13), C(13) and C(16) deviate most from the plane.

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References

CARRELL, H. L. (1976). VIEW. Institute for Cancer Research, The Fox Chase Cancer Center, Philadelphia, PA 19111, USA.

- CARRELL, H. L., SHIEH, H.-S. & TAKUSAGAWA, F. (1981). The Crystallographic Program Library of the Institute for Cancer Research, Fox Chase Cancer Center, Philadelphia, PA 19111, USA.
- DAVIES, E. K. (1983). The CHEMGRAF suite. Oxford Univ. Computing Laboratory, Oxford, England.
- FOSTER, R. (1969). Organic Charge Transfer Complexes. London: Academic Press.
- FOSTER, R., IBALL, J., SCRIMGEOUR, S. N. & WILLIAMS, B. C. (1976). J. Chem. Soc. Perkin Trans. 2, pp. 682–685.
- HERBSTEIN, F. H. (1971). Perspectives in Structural Chemistry, edited by J. D. DUNITZ & J. A. IBERS, pp. 166–395. New York: Wiley.
- JOHNSON, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
- JONES, T. A. (1978). J. Appl. Cryst. 11, 268-272.
- MAYOH, B. & PROUT, C. K. (1972). J. Chem. Soc. Faraday Trans. 2, 68, pp. 1072–1082.
- MUNNOCH, P. J. & WRIGHT, J. D. (1974). J. Chem. Soc. Perkin Trans. 2, pp. 1397–1400.
- MUNNOCH, P. J. & WRIGHT, J. D. (1975). J. Chem. Soc. Perkin Trans. 2, pp. 1071–1074.
- PROUT, C. K. & WRIGHT, J. D. (1968). Angew. Chem. Int. Ed. Engl. 7, 659–667.

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Structure of 4-(5-Cyano-2-furyl)-2,6-dimethyl-1,4-dihydropyridine-3,5-dicarbonitrile

By Jozef Kožíšek

Department of Inorganic Chemistry, Faculty of Chemical Technology, Radlinského 9, Slovak Technical University, CS - 812 37 Bratislava, Czechoslovakia

HELMUT PAULUS

Institut für Physikalische Chemie, Fachgebiet Strukturforschung, Technical University Darmstadt, Petersenstrasse 15, D - 6100 Darmstadt, Germany

and Štefan Marchalín and Dušan Ilavský

Department of Organic Chemistry, Faculty of Chemical Technology, Radlinského 9, Slovak Technical University, CS - 812 37 Bratislava, Czechoslovakia

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Abstract. $C_{14}H_{10}N_4O$, $M_r = 250.26$, monoclinic, $P2_1/c$, a = 13.293 (3), b = 8.099 (2), c = 13.815 (3) Å, $\beta = 120.474 (5)^{\circ}, V = 1281.9 (5) \text{ Å}^3, Z = 4, D_m =$ $D_x = 1.297 \text{ Mg m}^{-3}$, 1.26 (2), λ (Mo K α) = $\mu = 0.08 \text{ mm}^{-1}, \quad F(000) = 520,$ 0.71069 Å, T =299 K, final R = 0.0442 for 2081 unique reflections with $F > 2\sigma(F)$. The crystal structure consists of single molecules. The substituted 1,4-dihydropyridine ring has a flat boat conformation with the N(1) and C(4) atoms displaced by 0.100(1) and 0.141(1)Å, respectively, from the plane through the other four C atoms which defines the base of the boat. The plane of the furyl ring makes an angle of 76.7 (1)° with the

plane of the base of the boat. No significant intermolecular interactions are observed.

Introduction. Numerous newly synthesized 1,4dihydropyridines have been studied as models of NADH coenzymes in 'hydrogen-transfer' reactions (Eisner & Kuthan, 1972). In order for a 1,4dihydropyridine to act as a reducing agent, the H atom in the 4-position [H(3)] and that at the N atom should assume an antiperiplanar arrangement.

X-ray studies on 1,4-dihydropyridines have shown a great dependence of the geometry of the skeleton on the attached substituents, especially those in the 4

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position. Thus, (I) has a planar structure (Lenstra, Petit, Dommisse & Alderweireldt, 1979), (II) has a boat conformation (Fossheim, Svarteng, Mostad, Roemming, Shefter & Triggle, 1982) while (III) is a half-boat (Huml, Ječný, Kutschabsky & Reck, 1979).



We have studied the crystal structure of 4-(5cyano-2-furyl)-2,6-dimethyl-1,4-dihydropyridine-3,5dicarbonitrile (IV) prepared by a modified Hantzsch procedure starting from 5-formyl-2-furancarbonitrile and 3-amino-2-butenenitrile (Ilavský, Marchalín, Kováč, Henry-Basch & Morvan, 1987).

Experimental. A colourless well shaped crystal of dimensions $0.35 \times 0.48 \times 0.55$ mm was used. D_m was measured by flotation. Accurate unit-cell parameters were obtained by a least-squares fit of 42 reflections $(41.1 < 2\theta < 48.0^{\circ})$. The intensities were recorded on a Stoe–Siemens AED diffractometer for $3 < 2\theta < 50^{\circ}$ using graphite-monochromated Mo $K\alpha$ radiation and $\omega/2\theta$ scan mode. Index range was h 0/15, k -9/4, l - 16/14. Three standard reflections (020, 323, 323) were measured every 90 min; no significant systematic fluctuation was found. 3621 reflections were measured, 2250 unique, of which 2082 with F_{o} $> 2\sigma(F_{o})$ were considered observed; $R_{int} = 0.0105$ for 1184 unique reflections. The structure was solved by direct methods using SIR88 (Burla, Camalli, Cascarano, Giacovazzo, Polidori, Spagna & Viterbo, 1989) and refined on F by least squares with SHELX76 (Sheldrick, 1976). A difference map of the partially refined structure gave the positions of all H atoms. Refinement continued on all 213 positional and thermal parameters, anisotropic for non-H atoms and isotropic for H atoms. In the final cycle, R = 0.0442and wR = 0.0357 for 2081 observed reflections (one of the strongest reflections, 310, was considered to suffer from extinction and was excluded from the refinement), $w^{-1} = k\sigma^2(F)$, k = 3.1868; max. $\Delta/\sigma =$ 0.056; max. and min. heights in the final $\Delta \rho$ map 0.059 and $-0.066 \text{ e} \text{ Å}^{-3}$. No absorption correction was applied. Secondary-extinction correction refined

Table 1. Final atomic coordinates ($\times 10^4$) and equivalent isotropic thermal parameters $(Å^2 \times 10^3)^2$ with e.s.d.'s in parentheses

U_{eq} is defined as one	third of the trace	of the	orthogonalized		
U_{ij} tensor.					

	x	У	z	$U_{ m eq}$
O(1)	6760 (1)	618 (2)	6532 (1)	494 (6)
C(1)	5744 (2)	1446 (3)	6249 (2)	555 (10)
C(2)	5455 (2)	2460 (3)	5387 (2)	729 (12)
C(3)	6335 (2)	2274 (3)	5097 (2)	659 (10)
C(4)	7106 (1)	1164 (2)	5806 (1)	432 (8)
C(5)	5236 (2)	1079 (3)	6919 (2)	666 (11)
C(6)	8200 (1)	398 (2)	5955 (1)	412 (8)
C(7)	8692 (1)	1405 (2)	5362 (1)	416 (8)
C(8)	8540 (1)	964 (2)	4355 (1)	411 (9)
C(9)	7856 (1)	- 1728 (2)	4490 (1)	411 (9)
C(10)	7979 (1)	- 1353 (2)	5498 (1)	397 (7)
C(11)	7836 (2)	- 2600 (2)	6143 (1)	480 (9)
C(12)	7550 (2)	- 3394 (3)	3958 (2)	567 (12)
C(13)	8899 (2)	1979 (3)	3678 (2)	552 (11)
C(14)	8301 (2)	2893 (2)	5892 (1)	506 (9)
N(1)	8052 (1)	- 530 (2)	3902 (1)	453 (8)
N(2)	4848 (2)	786 (3)	7468 (1)	916 (12)
N(3)	9796 (2)	4070 (2)	6336 (1)	734 (9)
N(4)	7709 (2)	- 3558 (2)	6680 (1)	688 (10)

to 1.12 (2) × 10⁻⁶ [F_c multiplied by $(1 - xF_c^2)\sin\theta$]. The scattering factors for neutral atoms were taken from International Tables for X-ray Crystallography (1974, Vol. IV). No significant correlations among refined parameters were observed.

Discussion. Final atomic coordinates and equivalent isotropic temperature factors are given in Table 1* and bond lengths and angles (Nardelli, 1983) in Table 2. The molecular structure with atom numbering is shown in Fig. 1 (ORTEPII; Johnson, 1971).

The structure is formed by single molecules linked via hydrogen bonds (Table 2). The furyl ring is planar within experimental error. The dihydropyridine ring adopts a flattened boat conformation. The C(6), C(7), C(10) plane makes an angle of 17.9 (1)° with the C(7), C(8), C(9), C(10) plane, and the C(8), N(1), C(9) plane forms an angle of $171.3 (1)^{\circ}$ with the C(7), C(8), C(9), C(10) plane. The furyl ring plane forms an angle of 76.7 (1) $^{\circ}$ with plane C(7), C(8), C(9), C(10), thus confirming the $\sigma - \pi$ interaction of the p orbitals of 1,4-dihydropyridine and the σ bond C(6)(sp^3)—C(4)(sp^2). The interaction also manifests itself through the significant bathochromic shift of the longest-wavelength UV maximum. Bond distances and valence angles are normal. The 1,4-dihydropyridine cyano groups are equivalent and are involved in hydrogen bonds

^{*} Lists of structure factors, anisotropic thermal parameters and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 55732 (16 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: MU0297]

Table 2.	Selected in	iteratomi	c distance	es (Å),	angles	(°)
and hvdr	ogen bond	s (Å. °) и	vith e.s.d.	's in pa	irenthes	es

			-
O(1)—C(1)	1.3751 (30)	C(7)—C(8)	1.3486 (21)
O(1)-C(4)	1.3715 (23)	C(7)—C(14)	1.4292 (22)
C(1)—C(2)	1.3327 (37)	C(8)—C(13)	1.4936 (35)
C(1)-C(5)	1.4266 (47)	C(8)—N(1)	1.3656 (21)
C(2)—C(3)	1.4237 (46)	C(9)—C(10)	1.3518 (21)
C(3)-C(4)	1.3416 (25)	C(9)—C(12)	1.4910 (29)
C(4)—C(6)	1.4962 (21)	C(9)—N(1)	1.3725 (23)
C(5)—N(2)	1.1377 (41)	C(10)-C(11)	1.4219 (25)
C(6)—C(7)	1.5190 (23)	C(11)—N(4)	1.1436 (28)
C(6)—C(10)	1.5192 (23)	C(14)—N(3)	1.1439 (23)
C(1)-O(1)-C(4)	106.3 (2)	C(6)-C(7)-C(8	b) 122.3 (1)
O(1) - C(1) - C(5)	115.7 (2)	C(8)-C(7)-C(1	4) 120.1 (1)
O(1)-C(1)-C(2)	110.6 (3)	C(7) - C(8) - N(1)	119.7 (1)
C(2) - C(1) - C(5)	133.7 (3)	C(7) - C(8) - C(1)	3) 124.9 (2)
C(1) - C(2) - C(3)	106.3 (2)	C(13)-C(8)-N	(1) 115.4 (1)
C(2) - C(3) - C(4)	107.3 (2)	C(12)-C(9)-N	(1) 115.7 (2)
O(1) - C(4) - C(3)	109.6 (2)	C(10)-C(9)-N	(1) 119.4 (1)
C(3) - C(4) - C(6)	135.1 (2)	C(10)—C(9)—C((12) 124.9 (2)
O(1) - C(4) - C(6)	115.4 (1)	C(6)-C(10)-C((9) 122.3 (1)
C(1) - C(5) - N(2)	178.9 (3)	C(9)-C(10)-C((11) 120.1 (1)
C(4) - C(6) - C(10)	111.1 (1)	C(6)-C(10)-C((11) 117.5 (1)
C(4) - C(6) - C(7)	111.0 (1)	C(10)-C(11)-N	N(4) 177.4 (2)
C(7)—C(6)—C(10)	108.8 (1)	C(7)-C(14)-N	(3) 178.7 (1)
C(6)—C(7)—C(14)	117.6 (1)	C(8)—N(1)—C(9	P) 122.9 (1)
D—H···A	D…A	H…A .	D—H…A
$N(1) - H(4) - N(4^{i})$	2.959 (2)	2.10 (2)	177 (2)
C(13)-H(8)O(1") 3.480 (2)	2.72 (2)	138 (2)
C(6)—H(3)····N(3 ⁱⁱⁱ)	3.485 (2)	2.57 (1)	158 (1)
Symmetry code: (iii) $2 - x$, $-\frac{1}{2} + \frac{1}{2}$	(i) $x, -\frac{1}{2} - y,$ $y, \frac{3}{2} - z.$	$-\frac{1}{2}+z$; (ii) x	$, \frac{1}{2} - y, -\frac{1}{2} +$

(Table 2). Double-bond lengths in the furyl ring differ very slightly [C(1)=C(2) = 1.333 (4) and C(3)=C(4) = 1.342 (3) Å] owing to the conjugation in the N(2)(sp)-C(5)(sp)-C(1)(sp²)-C(2)(sp²)-C(3)(sp²)-C(4)(sp²) unit. The replacement of 4methylpyridine in position 5 of the furan ring (Lokaj, Vrábel, Sivý, Kettmann, Ilavský & Ječný, 1991) by the cyano group has resulted in increased conjugation, leading to a decrease in the C(2)-C(3) bond length from 1.441 (3) to 1.424 (5) Å.

Atomic thermal motion increases significantly with distance from the molecular centre. The U_{eq} values range from <0.042 Å² for dihydropyridine to



Fig. 1. The molecular structure of the title compound showing the atom numbering.

0.073 Å² in the furyl ring and up to 0.092 (1) Å² for the outer N(2) atom. The ratio of the shortest to longest principal axis increases likewise, with a maximum asymmetry of 2.98 for N(4).

References

- BURLA, M. C., CAMALLI, M., CASCARANO, G., GIACOVAZZO, C., POLIDORI, G., SPAGNA, R. & VITERBO, D. (1989). J. Appl. Cryst. 22, 389-393.
- EISNER, U. & KUTHAN, J. (1972). Chem. Rev. 72, 1-42.
- Fossheim, R., Svarteng, K., Mostad, A., Roemming, C., Shefter, E. & Triggle, D. J. (1982). *J. Med. Chem.* 25, 126–131.
- HUML, K., JEČNÝ, J., KUTSCHABSKY, L. & RECK, G. (1979). Acta Cryst. B35, 444-447.
- ILAVSKÝ, D., MARCHALÍN, S., KOVÁČ, J., HENRY-BASCH, E. & MORVAN, C. (1987). Chem. Papers, 41, 793–801.
- JOHNSON, C. K. (1971). ORTEPII. Report ORNL-3794, revised. Oak Ridge National Laboratory, Tennessee, USA.
- LENSTRA, A. T. H., PETIT, G. H., DOMMISSE, R. A. & ALDERWEIRELDT, F. C. (1979). Bull. Soc. Chim. Belg. 88, 133-141.
- LOKAJ, J., VRÁBEL, V., SIVÝ, P., KETTMANN, V., ILAVSKÝ, D. & JEČNÝ, J. (1991). Acta Cryst. C47, 886–888.
- NARDELLI, M. (1983). Comput. Chem. 7, 95-98.
- SHELDRICK, G. M. (1976). SHELX76. Program for crystal structure determination. Univ. of Cambridge, England.

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Structure of 12b-Hydroxy-5-methyl-1,2,3,4,4a,5,6,12b-octahydro-12*H*benzo[*a*]xanthen-12-one

BY ROY M. LETCHER,* KUNG-KAI CHEUNG* AND TAI-YUEN YUE

Department of Chemistry, University of Hong Kong, Bonham Road, Hong Kong

(Received 15 May 1992; accepted 26 August 1992)

Abstract. $C_{18}H_{20}O_3$, $M_r = 284.35$, triclinic, $P\bar{1}$, a = 7.505 (2), b = 8.805 (2), c = 11.443 (4) Å, $\alpha =$

* Authors to whom correspondence should be addressed.

101.53 (3), $\beta = 99.57$ (3), $\gamma = 95.43$ (2)°, V = 724.3 (9) Å³, Z = 2, $D_x = 1.299$ g cm⁻³, λ (Mo K α) = 0.71073 Å, $\mu = 0.816$ cm⁻¹, F(000) = 304, T = 296 K, R = 0.045 for 1900 independent reflections.

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